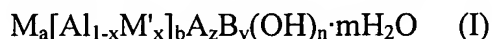


### AMENDMENTS TO THE CLAIMS

This listing of claims will replace all prior versions, and listings, of claims in the present application.

#### Listing of Claims:

1. **(Currently Amended)** Organic acid anion containing aluminum salt hydroxide particles represented by the following general formula (I):



(wherein M is at least one cation selected from the group consisting of  $Na^+$ ,  $K^+$ ,  $[Li]^+$  and  $NH_4^+$  and  $H_3O^+$ ,  $M'$  is at least one metal cation selected from the group consisting of  $Cu^{2+}$ ,  $Zn^{2+}$ ,  $Ni^{2+}$ ,  $Sn^{4+}$ ,  $Zr^{4+}$ ,  $Fe^{2+}$ ,  $Fe^{3+}$  and  $Ti^{4+}$ , A is at least one organic acid anion based on an organic acid selected from the group consisting of (i) an organic carboxylic acid having 2 to 10 carbon atoms and 1 to 4 carboxyl groups and (ii) an organic oxycarboxylic acid having 2 to 10 carbon atoms and 1 to 4 carboxyl groups, B is at least one inorganic acid anion selected from the group consisting of a sulfate ion, a phosphate ion and a nitrate ion, and a, b, m, n, x, y and z satisfy  $0.7 \leq a \leq 1.35$ ,  $2.7 \leq b \leq 3.3$ ,  $0 \leq m \leq 5$ ,  $4 \leq n \leq 7$ ,  $0 \leq x \leq 0.6$ ,  $1.7 \leq y \leq 2.4$ , and  $0.001 \leq z \leq 0.5$ , respectively.)

2. **(Original)** The particles according to claim 1, which are represented by the formula (I) wherein a satisfies  $0.9 \leq a \leq 1.2$ .

3. **(Original)** The particles according to claim 1, which are represented by the formula (I) wherein b satisfies  $2.8 \leq b \leq 3.2$ .

4. **(Original)** The particles according to claim 1, which are represented by the formula (I) wherein m satisfies  $0 \leq m \leq 2$ .

5. **(Original)** The particles according to claim 1, which are represented by the formula (I) wherein  $n$  satisfies  $5 \leq n \leq 6.5$ .

6. **(Original)** The particles according to claim 1, which are represented by the formula (I) wherein  $x$  satisfies  $0 \leq x \leq 0.3$ .

7. **(Original)** The particles according to claim 1, which are represented by the formula (I) wherein  $y$  satisfies  $1.8 \leq y \leq 2.2$ .

8. **(Original)** The particles according to claim 1, which are represented by the formula (I) wherein  $z$  satisfies  $0.01 \leq z \leq 0.4$ .

9. **(Previously Presented)** The particles according to claim 1, wherein the organic acid anion (A) in the formula (I) is at least one selected from anions based on an oxalic acid, a citric acid, a citrate, a tartaric acid, a tartrate, a DL-malic acid, a gallic acid, a DL-glyceric acid and an L-lactic acid.

10-12. **(Cancelled)**

13. **(Original)** The particles according to claim 1, wherein  $D_{25}$  and  $D_{75}$  satisfy  $1 < D_{75}/D_{25} < 1.8$  when particle diameters at 25% and 75% values of cumulative particle size distribution curve measured by a laser diffraction method are represented by  $D_{25}$  and  $D_{75}$ , respectively.

14. **(Original)** The particles according to claim 1, which are in the shape of grains, pairs, rectangular parallelepiped, disks (go stones), hexagonal plates, rice grains or cylinders.

15. **(Original)** The particles according to claim 1, having an average particle diameter of 0.1 to 10  $\mu\text{m}$ .

16. **(Withdrawn)** A burned product obtained by burning the organic acid anion containing aluminum salt hydroxide particles of claim 1 at 300 to 1,000°C.

17. **(Original)** The particles according to claim 1, which carry a hydrolysate of a salt of at least one metal selected from the group consisting of Cu, Zn, Ni, Sn, Zr, Fe and Ti, on the surfaces thereof.

18. **(Withdrawn)** The alunite type compound particles of claim 1, having surfaces thereof treated with at least one surface treating agent selected from the group consisting of a higher fatty acid, an anionic surfactant, a phosphoric ester, a coupling agent and an ester of a polyhydric alcohol and a fatty acid.

19. **(Canceled)**

20. **(Previously Presented)** The method according to claim 30, wherein the inorganic salt is aluminum sulfate.

21-23. **(Cancelled)**

24. **(Previously Presented)** The method according to claim 30, wherein the heating reaction is carried out at 90 to 250°C.

25. **(Withdrawn)** A resin additive comprising the organic acid anion containing aluminum salt hydroxide particles of claim 1.

26. **(Withdrawn)** A resin composition containing the resin additive of claim 25.

27. **(Withdrawn)** An adsorbent composition containing the organic acid anion containing aluminum salt hydroxide particles of claim 1.

28. **(Withdrawn)** A dye carrier containing the organic acid anion containing aluminum salt hydroxide particles of claim 1.

29. **(Withdrawn)** An ultraviolet absorber containing the organic acid anion containing aluminum salt hydroxide particles of claim 1.

30. **(Currently Amended)** A method for producing organic acid anion containing aluminum salt hydroxide particles of claim 1, which comprises adding ~~an alkali hydroxide solution~~ a solution of a hydroxide of an ion selected from the group consisting of  $\text{Na}^+$ ,  $\text{K}^+$ [[,]] and  $\text{NH}_4^+$  ~~and  $\text{H}_3\text{O}^+$~~  to a mixed solution comprising an organic acid or organic acid salt selected from the group consisting of (i) an organic carboxylic acid having 2 to 10 carbon atoms and 1 to 4 carboxyl groups, (ii) an organic oxycarboxylic acid having 2 to 10 carbon atoms and 1 to 4 carboxyl groups, and (iii) salts thereof, an inorganic salt of  $\text{Al}^{3+}$  selected from the group consisting of an aluminum sulfate, an aluminum phosphate and an aluminum nitrate, and a sulfate or nitrate of at least one member selected from the group consisting of  $\text{Na}^+$ ,  $\text{K}^+$ [[,]] and  $\text{NH}_4^+$  ~~and  $\text{H}_3\text{O}^+$~~  to cause a heating reaction and produce the organic acid anion containing aluminum salt hydroxide particles of claim 1.

31. **(Previously Presented)** The method according to claim 30, wherein the mixed solution further comprises an inorganic salt of at least one cation selected from the group consisting of  $\text{Cu}^{2+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Sn}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Fe}^{3+}$  and  $\text{Ti}^{4+}$ .

32-35. **(Cancelled)**

36. **(Previously Presented)** The method according to claim 31, wherein the heating reaction is carried out at 90 to 250°C.